

9E Reactions of metals and metal compounds

complete the sentence

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Place the right word or words to complete the sentence:



Click mouse to reveal answer

two

zinc

metals

word

atoms

conserved

formula

oxide

acid

- A compound contains ____ more different elements joined together.
- A _____ tells you the type and number of atoms present in a compound.
- During a reaction, _____ join other atoms or molecules to form new products.
- Reactive _____ can easily combine with oxygen, water and lab acids.
- The formula for magnesium _____ is MgO.
- When you add a reactive metal to _____, hydrogen gas is liberated from the acid.
- Reactions can be described using a _____ or symbol equation.
- Mass is always _____ during a chemical reaction.
- Iron is less reactive than _____, which in turn is less reactive than potassium.

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- A compound contains two more different elements joined together.
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- During a reaction, atoms join other atoms or molecules to form new products.
- Reactive metals can easily combine with oxygen, water and lab acids.
- The formula for magnesium oxide is MgO.
- When you add a reactive metal to acid, hydrogen gas is liberated from the acid.
- Reactions can be described using a word or symbol equation.
- Mass is always conserved during a chemical reaction.
- Iron is less reactive than zinc, which in turn is less reactive than potassium.